



**Applied Organic Chemistry 513 CHM**

**الإجابة النموذجية لامتحان مادة الكيمياء العضوية التطبيقية 513 ك**

**ورقة امتحانية كاملة (ساعتان)**

**الفرقة : دبلومه كيمياء تحليليه**

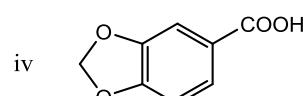
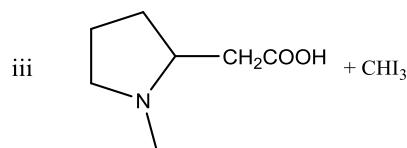
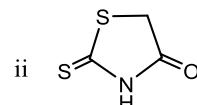
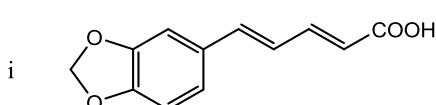
**التاريخ : الاثنين 20 / 1 / 2020**

**الممتحن : أ.د/ محمد سيد عبد الرحمن سيد بحallo**

**قسم : الكيمياء**

**كلية : العلوم**

**1-**



**2-** Antiviral, anticancer, aldose reductase inhibitor, antibacterial, antifungal, antidiabetic and pesticidal,

**3-**

i- Elemental analysis and MW det. → MF is  $C_{10}H_{15}NO$

ii- Reaction with  $HNO_2$  → -Ve, tertiary amine

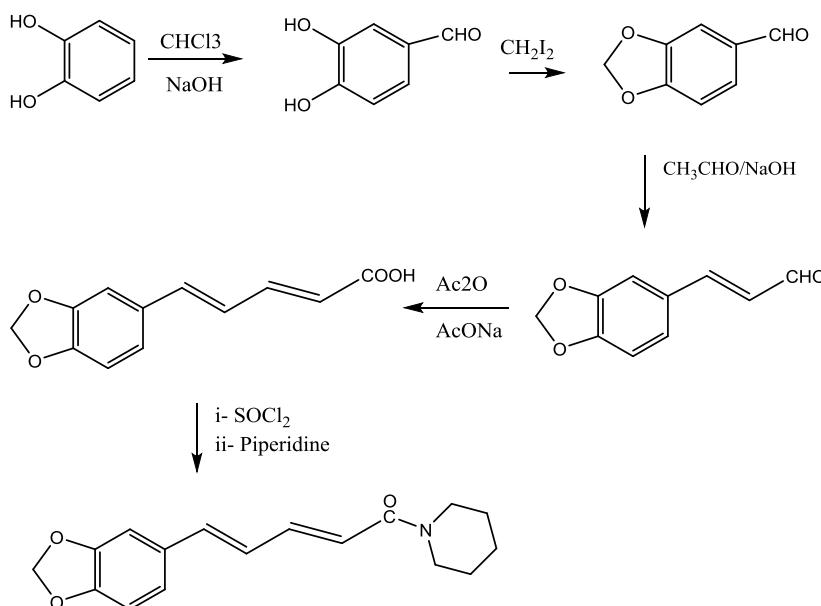
iii- Reaction with benzene sulphonyl chloride → -Ve, tertiary amine

iv- It reacts with one mole acetic anhydride  $\longrightarrow$  it has one hydroxyl group

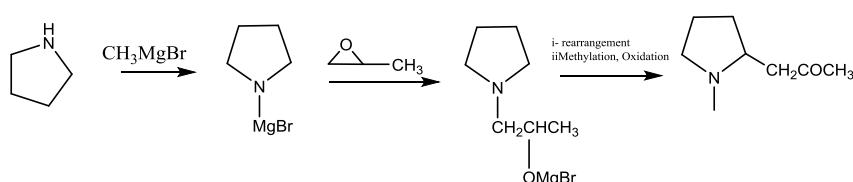
v- It reacts with  $\text{FeCl}_3$  and give dark color  $\longrightarrow$  phenolic OH

vi- Oxidation give p-anisic acid  $\longrightarrow$  mono side chain

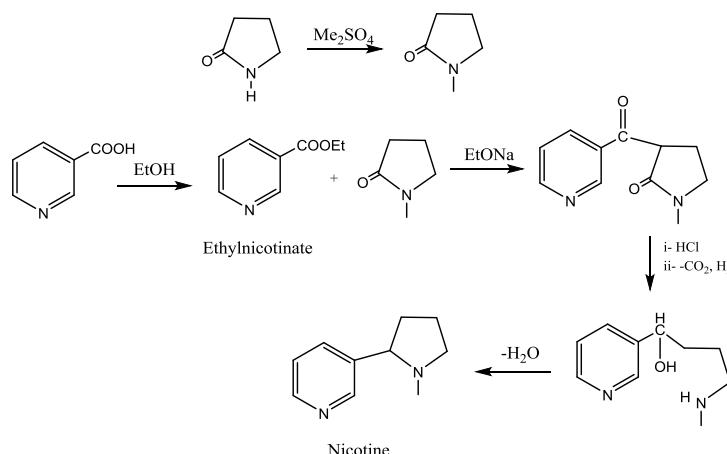
4- i-



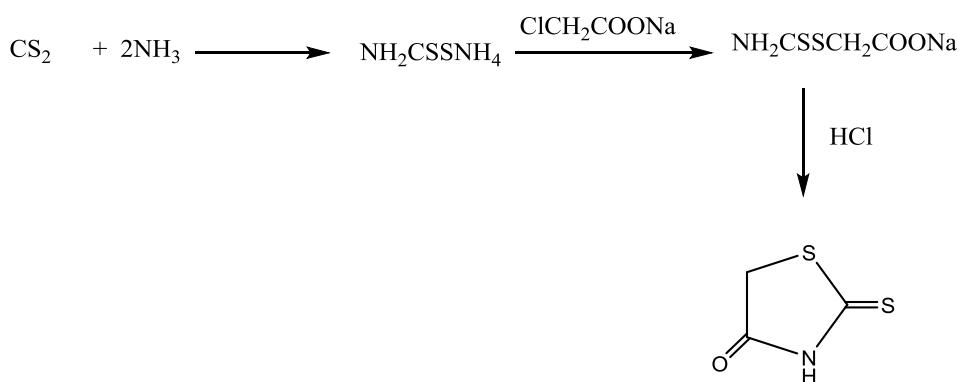
ii-



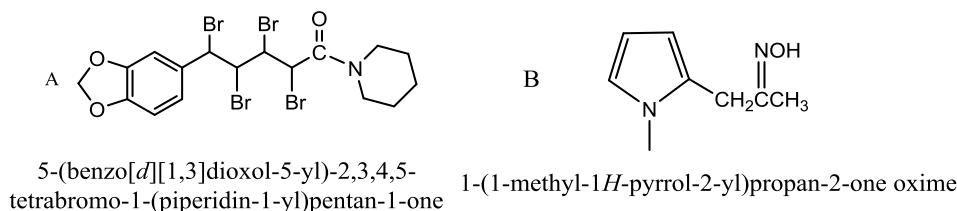
iii-



iv-



5-



6-

- i- 3-(1-methylpyrrolidin-2-yl)pyridine
- ii-
- 1- Elemental analysis and MW det.  $\longrightarrow$  MF is  $\text{C}_{10}\text{H}_{14}\text{N}_2$
- 2- Reaction with  $\text{HNO}_2$   $\longrightarrow$  -Ve, tertiary amine
- 3- Reaction with benzene sulphonyl chloride  $\longrightarrow$  -Ve, tertiary amine
- 4- Heating with  $\text{ZnCl}_2$   $\longrightarrow$  mixture of pyridine, pyrrole and methyl amine  
(methyl group at N )
- 5- Oxidation of nicotine gives nicotinic acid  $\longrightarrow$  side chain of pyridine at position 3
- 6- Reduction by hydrogen consume only  $3\text{H}_2$   $\longrightarrow$  saturation at pyrrole